Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 288 KMean σ (C–C) = 0.003 Å R factor = 0.052 wR factor = 0.124 Data-to-parameter ratio = 14.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

In the title compound, $C_{16}H_9N_3$, the phenyl ring makes a dihedral angle of 70.7 (3)° with the tricyanophenyl ring and there are intermolecular C-H···N interactions in the crystal structure.

5-Benzylbenzene-1,2,4-tricarbonitrile

Received 12 September 2005 Accepted 30 September 2005 Online 5 October 2005

Comment

Photo-induced electron-transfer (PET) reactions of cyanoarenes with alkenes have been the subject of active research (Mangion & Arnold, 2002). 1,2,4,5-Tetracyanobenzene (TCNB) is the strongest electron acceptor of all cyanoarenes (Mattes & Farid, 1982). In our ongoing research work on PET reactions between TCNB and different substrates, we have prepared the title compound, (I), which was obtained from the PET reaction of TCNB with styrene in a polar solvent mixture (acetonitrile–water = 85:15 v/v). As part of this study, we have undertaken an X-ray crystallographic analysis of (I) in order to elucidate its conformation and configuration.



The bond lengths and angles in (I) are in good agreement with expected values, except for the $C_{ar}-C(\equiv N)$ bond lengths [1.434 (3)–1.440 (3) Å], which are slightly longer than the typical $Csp-Csp^2$ bond distance (Allen *et al.*, 1987). These $C_{ar}-C_{CN}$ bond lengths are comparable with those observed in the related compounds 1,2,4,5-benzenetetracarbonitrile–acridine (1/1) [1.444 (7) Å; Toupet *et al.*, 1989], 4phenyl-1-(2,4,5-tricyanophenyl)-1,2,3,4-tetrahydronaphthalene [1.442 (4)–1.446 (4) Å; Zhang *et al.*, 2002] and 5-(1,4dimethyl-4-phenyl-1,2,3,4-tetrahydro-1-naphthyl)-1,2,4-benzenetricarbonitrile [1.440 (4)–1.446 (4) Å; Usman *et al.*, 2002]. The dihedral angle formed between the phenyl and the tricyanophenyl rings is 70.7 (3)°, with atom C7 lying on the intersection of the two planes. There are intermolecular C– H···N interactions in the crystal structure (Table 2).

Experimental

The title compound, (I), was prepared by the photo-induced reaction of 1,2,4,5-tetracyanobenzene with 1.5 equiv. of styrene in a polar solvent mixture (acetonitrile–water = 85:15 v/v), irradiated by light of wavelength longer than 300 nm for 120 h. It was isolated by column

Acta Cryst. (2005). E61, o3543-o3544

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chromatography of the reaction mixture after evaporation of the solvent on silica gel. Single crystals of (I) were obtained by slow evaporation of a petroleum ether ethyl-acetate (3:1 v/v) solution (yield 25%).

Crystal data

 $\begin{array}{l} C_{16}H_9N_3\\ M_r = 243.26\\ Monoclinic, P2_1/c\\ a = 11.289 \ (2) \ \text{\AA}\\ b = 8.6380 \ (17) \ \text{\AA}\\ c = 14.193 \ (3) \ \text{\AA}\\ \beta = 113.42 \ (3)^{\circ}\\ V = 1270.0 \ (4) \ \text{\AA}^3\\ Z = 4 \end{array}$

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (XCAD4; Harms & Wocadlo, 1995) $T_{\min} = 0.954$, $T_{\max} = 0.977$ 2622 measured reflections 2493 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.052$ $wR(F^2) = 0.124$ S = 1.002493 reflections 174 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

C4-C7	1.505 (3)	C11-C16	1.434 (3)
C7-C8	1.506 (3)	C12-C15	1.439 (3)
C9-C14	1.440 (3)		()
C4-C7-C8	112.26 (17)		

Table 2

Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
$C1 - H1 \cdots N2^i$	0.93	2.73	3.621 (3)	161
Symmetry code: (i)	$r = 1 = v \pm \frac{5}{7}$	_ 1		

Symmetry code: (i) $x - 1, -y + \frac{5}{2}, z - \frac{1}{2}$.

 $D_x = 1.272 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 25 reflections $\theta = 10-13^\circ$ $\mu = 0.08 \text{ mm}^{-1}$ T = 288 (2) KBlock, colourless $0.40 \times 0.33 \times 0.30 \text{ mm}$

1560 reflections with $I > 2\sigma(I)$ $R_{int} = 0.027$ $\theta_{max} = 26.0^{\circ}$ $h = 0 \rightarrow 13$ $k = 0 \rightarrow 10$ $l = -16 \rightarrow 16$ 3 standard reflections every 200 reflections intensity decay: none

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0301P)^{2} + 0.8P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.006$ $\Delta\rho_{max} = 0.20 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.25 \text{ e} \text{ Å}^{-3}$





The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C–H distances in the range 0.93–0.97 Å and with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

This work was supported by the National Natural Science Foundation of China (NSFC grant No. 20272024).

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