Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Yan-Guang Wang, Hui-Ying An, Zhi-Feng Lu, Lin Wu and Jian-Hua Xu*

Department of Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

Correspondence e-mail: xujh@nju.edu.cn

## Key indicators

Single-crystal X-ray study
$T=288 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.052$
$w R$ factor $=0.124$
Data-to-parameter ratio $=14.3$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2005 International Union of Crystallography Printed in Great Britain - all rights reserved

## 5-Benzylbenzene-1,2,4-tricarbonitrile

In the title compound, $\mathrm{C}_{16} \mathrm{H}_{9} \mathrm{~N}_{3}$, the phenyl ring makes a dihedral angle of 70.7 (3) ${ }^{\circ}$ with the tricyanophenyl ring and there are intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ interactions in the crystal structure.

## Comment

Photo-induced electron-transfer (PET) reactions of cyanoarenes with alkenes have been the subject of active research (Mangion \& Arnold, 2002). 1,2,4,5-Tetracyanobenzene (TCNB) is the strongest electron acceptor of all cyanoarenes (Mattes \& Farid, 1982). In our ongoing research work on PET reactions between TCNB and different substrates, we have prepared the title compound, (I), which was obtained from the PET reaction of TCNB with styrene in a polar solvent mixture (acetonitrile-water $=85: 15 \mathrm{v} / \mathrm{v}$ ). As part of this study, we have undertaken an X-ray crystallographic analysis of (I) in order to elucidate its conformation and configuration.

(I)

The bond lengths and angles in (I) are in good agreement with expected values, except for the $\mathrm{C}_{\mathrm{ar}}-\mathrm{C}(\equiv \mathrm{N})$ bond lengths $[1.434$ (3) -1.440 (3) Å], which are slightly longer than the typical Csp-Csp ${ }^{2}$ bond distance (Allen et al., 1987). These $\mathrm{C}_{\mathrm{ar}}-\mathrm{C}_{\mathrm{CN}}$ bond lengths are comparable with those observed in the related compounds 1,2,4,5-benzenetetracarbonitrile-acridine (1/1) [1.444 (7) £ं; Toupet et al., 1989], 4-phenyl-1-(2,4,5-tricyanophenyl)-1,2,3,4-tetrahydronaphthalene [1.442 (4)-1.446 (4) Aं; Zhang et al., 2002] and 5-(1,4-dimethyl-4-phenyl-1,2,3,4-tetrahydro-1-naphthyl)-1,2,4-benzenetricarbonitrile [1.440 (4)-1.446 (4) Å; Usman et al., 2002]. The dihedral angle formed between the phenyl and the tricyanophenyl rings is $70.7(3)^{\circ}$, with atom C 7 lying on the intersection of the two planes. There are intermolecular C$\mathrm{H} \cdots \mathrm{N}$ interactions in the crystal structure (Table 2).

## Experimental

The title compound, (I), was prepared by the photo-induced reaction of 1,2,4,5-tetracyanobenzene with 1.5 equiv. of styrene in a polar solvent mixture (acetonitrile-water $=85: 15 \mathrm{v} / \mathrm{v}$ ), irradiated by light of wavelength longer than 300 nm for 120 h . It was isolated by column

Received 12 September 2005 Accepted 30 September 2005 Online 5 October 2005
chromatography of the reaction mixture after evaporation of the solvent on silica gel. Single crystals of (I) were obtained by slow evaporation of a petroleum ether ethyl-acetate ( $3: 1 \mathrm{v} / \mathrm{v}$ ) solution (yield $25 \%$ ).

## Crystal data

$\mathrm{C}_{16} \mathrm{H}_{9} \mathrm{~N}_{3}$
$M_{r}=243.26$
Monoclinic, $P 2_{6} / c$
$a=11.289(2) \AA$
$b=8.6380(17) \AA$
$c=14.193(3) \AA$
$\beta=113.42(3)^{\circ}$
$V=1270.0(4) \AA^{3}$
$Z=4$
$D_{x}=1.272 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$M_{r}=243.26$
Monoclinic, $P 2_{1} / c$
Cell parameters from 25
reflections
$\theta=10-13^{\circ}$
$\mu=0.08 \mathrm{~mm}^{-1}$
$T=288$ (2) K
Block, colourless
$0.40 \times 0.33 \times 0.30 \mathrm{~mm}$
Data collection
Enraf-Nonius CAD-4
diffractometer
1560 reflections with $I>2 \sigma(I)$
$\omega / 2 \theta$ scans
$R_{\text {int }}=0.027$
Absorption correction: $\psi$ scan
( $X C A D 4$; Harms \& Wocadlo, 1995)
$T_{\text {min }}=0.954, T_{\text {max }}=0.977$
2622 measured reflections
2493 independent reflections
$\theta_{\text {max }}=26.0^{\circ}$
$h=0 \rightarrow 13$
$k=0 \rightarrow 10$
$l=-16 \rightarrow 16$
3 standard reflections every 200 reflections intensity decay: none

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.052$
$w R\left(F^{2}\right)=0.124$
$S=1.00$
2493 reflections
174 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0301 P)^{2}\right. \\
& \quad+0.8 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.006 \\
& \Delta \rho_{\max }=0.20 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.25 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{C} 4-\mathrm{C} 7$ | $1.505(3)$ | $\mathrm{C} 11-\mathrm{C} 16$ | $1.434(3)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{C} 7-\mathrm{C} 8$ | $1.506(3)$ | $\mathrm{C} 12-\mathrm{C} 15$ | $1.439(3)$ |
| $\mathrm{C} 9-\mathrm{C} 14$ | $1.440(3)$ |  |  |
| $\mathrm{C} 4-\mathrm{C} 7-\mathrm{C} 8$ | $112.26(17)$ |  |  |

Table 2
Hydrogen-bond geometry $\left(\AA^{\circ},^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 1-\mathrm{H} 1 \cdots \mathrm{~N} 2^{\mathrm{i}}$ | 0.93 | 2.73 | $3.621(3)$ | 161 |

Symmetry code: (i) $x-1,-y+\frac{5}{2}, z-\frac{1}{2}$.


Figure 1
The structure of (I), showing 50\% probability displacement ellipsoids and the atom-numbering scheme.

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-$ $0.97 \AA$ and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: XCAD4 (Harms \& Wocadlo, 1995); program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: $S H E L X T L$; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

This work was supported by the National Natural Science Foundation of China (NSFC grant No. 20272024).

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Enraf-Nonius (1989). CAD-4 Software. Version 5. Enraf-Nonius, Delft, The Netherlands.
Harms, K. \& Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
Mangion, D. \& Arnold, D. R. (2002). Acc. Chem. Res. 35, 297-304.
Mattes, S. L. \& Farid, S. (1982). Acc. Chem. Res. 15, 80-86.
Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Toupet, L., Minewicz, A. \& Ecolivet, C. (1989). Acta Cryst. C45, 1044-1047.
Usman, A., Razak, I. A., Fun, H.-K., Chantrapromma, S., Zhang, M. \& Xu, J.H. (2002). Acta Cryst. E58, o467-o468.

Zhang, M., Usman, A., Razak, I. A., Fun, H.-K., Chantrapromma, S. \& Xu, J.H. (2002). Acta Cryst. E58, o132-o133.

